

# Luminescence characteristics of $\text{PrCl}_3:\text{Ce}^{3+}$

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$\text{PrBr}_3:\text{Ce}^{3+}$  was recently introduced as a new fast scintillator [1]. Its scintillation decay curve is non-exponential. The optically excited 5d state of  $\text{Ce}^{3+}$  shows 6 ns decay time constant at room temperature (RT), which is faster than the 15 ns of  $\text{LaBr}_3:\text{Ce}^{3+}$  [2,3]. The decay time constant of the  $\text{Ce}^{3+}$  5d state of  $\text{PrBr}_3:\text{Ce}^{3+}$  at 10 K is 11 ns. The 5 ns shortening of the  $\text{Ce}^{3+}$  5d decay time when the temperature increases from 10 K to RT points to  $\text{Ce}^{3+}$  emission quenching, and the reason of the low light yield in  $\text{PrBr}_3:\text{Ce}^{3+}$  [1]. To understand the nature of this quenching, we studied the scintillation and luminescence properties of  $\text{PrCl}_3:\text{Ce}^{3+}$ .

$\text{PrCl}_3:\text{Ce}^{3+}$  crystallizes in the same structure as  $\text{PrBr}_3:\text{Ce}^{3+}$  ( $\text{UCl}_3$  type structure with space group  $\text{P6}_3/\text{m}$ ). The crystals are hygroscopic. Emission and excitation spectra were recorded at the SUPERLUMI station of HASYLAB. Details of this setup were described elsewhere [4].

Fig. 1a shows the emission spectra of  $\text{PrCl}_3:5\%\text{Ce}^{3+}$ . With a photon excitation at 4.43 eV and recorded at 10 K,  $\text{PrCl}_3:5\%\text{Ce}^{3+}$  exhibits d-f doublet emission of  $\text{Ce}^{3+}$  peaking at 3.43 and 3.67 eV, see dashed lines in Fig. 1a. Compared to  $\text{LaCl}_3:\text{Ce}^{3+}$ , the  $\text{Ce}^{3+}$  emission is 0.1 eV shifted to lower energy. When we excite at 5.74 eV,  $\text{Pr}^{3+}$  4f $\rightarrow$ 4f transition lines are also present in the spectrum. These lines originate from the  $^3\text{P}_0$  state. These lines were previously reported for undoped  $\text{PrCl}_3$  and  $\text{PrBr}_3$ . The absence of the emission from the  $^3\text{P}_1$  state has been attributed to the energy exchange with neighboring  $\text{Pr}^{3+}$  ions [5].

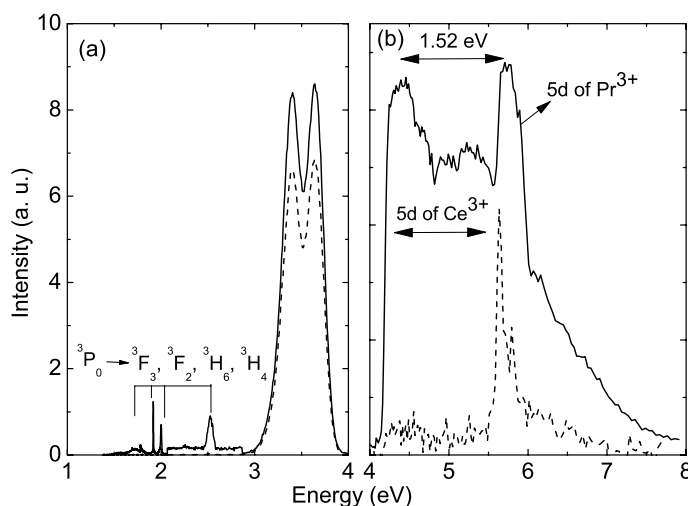


Figure 1: (a) Emission spectra of  $\text{PrCl}_3:5\%\text{Ce}^{3+}$  excited at 5.74 (solid line) and at 4.43 eV (dashed line) and (b) excitation spectra of  $\text{PrCl}_3:5\%\text{Ce}^{3+}$  monitoring 3.69 (solid line) and 2.53 eV (dashed line) emission. All spectra were recorded at 10 K.

The excitation spectra of  $\text{PrCl}_3:5\%\text{Ce}^{3+}$  are shown in Fig. 1b. The excitation spectrum monitoring  $\text{Ce}^{3+}$  emission at 3.69 eV shows several bands between 4.13 and 5.91 eV, see solid line in Fig. 1b. The bands between 4.13 and 5.39 eV are assigned to the interconfigurational  $\text{Ce}^{3+}[\text{Xe}]4\text{f}^1 \rightarrow [\text{Xe}]5\text{d}^1$  transitions. A different band at 5.74 eV also appears in the excitation spectrum monitoring the  $\text{Pr}^{3+}$   $^3\text{P}_0 \rightarrow ^3\text{H}_4$  transition at 2.53 eV, see dashed line in Fig. 1b. This band is assigned to the  $4\text{f} \rightarrow 5\text{d}_{\text{lowest}}$

transition of  $\text{Pr}^{3+}$ . The assignment is based on the estimation of the energy of the  $4f \rightarrow 5d_{\text{lowest}}$  transition of  $\text{Pr}^{3+}$  from that of  $4f \rightarrow 5d_{\text{lowest}}$  transition in  $\text{Ce}^{3+}$ . The  $\text{Pr}^{3+} 4f^1 5d^1$  bands are always about 1.52 eV higher in energy than those of  $\text{Ce}^{3+}$  [6].

Fig. 2 shows decay curves of  $\text{PrCl}_3:5\% \text{Ce}^{3+}$  monitoring  $\text{Ce}^{3+}$  emission of 3.69 eV excited via the 5d band of  $\text{Ce}^{3+}$  at 4.43 eV recorded at 10 K and RT. The scintillation decay curve of  $\text{PrCl}_3:5\% \text{Ce}^{3+}$  under  $^{137}\text{Cs}$  662 keV  $\gamma$ -ray excitation recorded at RT is shown in the inset. Details of the scintillation decay time setup are given elsewhere [1]. Both decay curves of  $\text{Ce}^{3+}$  emission excited via the 5d band of  $\text{Ce}^{3+}$  at 10 K and RT show a single exponential of 12 ns. Unlike  $\text{PrBr}_3:5\% \text{Ce}^{3+}$ ,  $\text{PrCl}_3:5\% \text{Ce}^{3+}$  does not show  $\text{Ce}^{3+}$  emission quenching between 10 K and RT. The scintillation decay curve shows two components of 17 and 230 ns. Their contributions to the total light yield are 80 and 20 %, respectively. The 5 ns difference between the 12 ns decay time of the 5d excited state of  $\text{Ce}^{3+}$  and the 17 ns fast component of the scintillation decay curve implies that the main transfer from the host to  $\text{Ce}^{3+}$  takes place in about 5 ns.

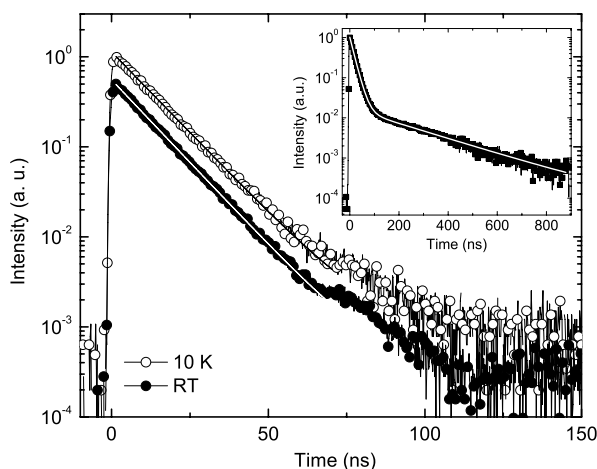


Figure 2: Decay curves of  $\text{PrCl}_3:5\% \text{Ce}^{3+}$  monitoring emission of 3.69 eV excited at 4.43 eV. Decay curves with empty circles (o) were recorded at 10 K whereas those with solid circles (●) were recorded at RT. The inset shows the scintillation decay curve under  $^{137}\text{Cs}$  662 keV  $\gamma$ -ray excitation recorded with single photon counting method at RT. Solid lines through the data are mono- and biexponential fits.

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